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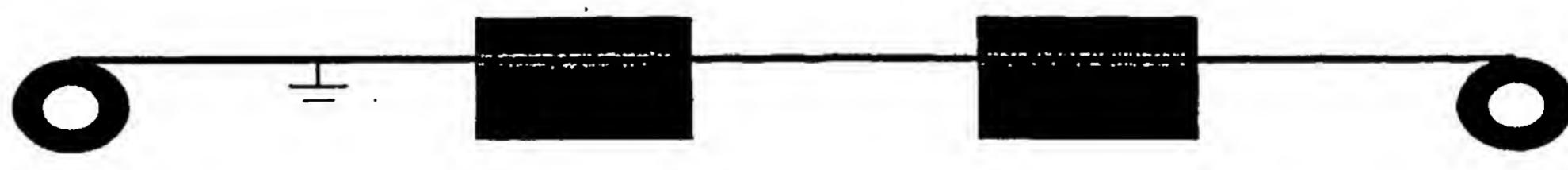
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(54) Title: ELECTROSTATIC IMPREGNATION OF POWDERS ON SUBSTRATES



Fiber Spool

Grounding Unit

Electrostatic Coater

Furnace/Oven

Take-Up

(57) Abstract: Electrostatic Impregnation Of Powders On Substrates The use of electrostatic impregnation to load materials such as binders and flavors onto substrates such as fibers and medical devices is disclosed. Substrates loaded with materials such as binders and flavors, wherein the materials are loaded on the substrates via electrostatic impregnation are also disclosed.

Electrostatic Impregnation Of Powders On Substrates

The present invention relates to the use of electrostatic impregnation to load materials such as binders and flavors onto substrates such as fibers and medical devices made from ceramics, metal alloys, and polymers. The invention also relates to substrates loaded with materials such as binders and flavors, wherein the materials are loaded into or on the substrates via electrostatic impregnation.

10

Many substrates are coated with materials such as polymers, polymeric binders, wax binders, flavors, and the like. For example, medical devices, such as stents, which are used in the human body, are frequently made of metal alloys. The stents require coating with a polymer or wax before use in the body. Another example of a substrate that is typically coated, e.g., with a polymer, a wax or the like, is dental floss.

20 Currently, dental floss has three main consumer needs that are not achieved in all products. These needs are (i) prevention/ minimization of fraying and shredding during use, (ii) easy insertion and sliding between tight teeth and (iii) gentleness to the gums. As used herein, 25 "fraying" means the separation of fibers by the stress placed on the floss during use between the teeth. As used herein, "shredding" means the breaking of fibers by the stress placed on the floss during use between the teeth. The minimization of fraying and shredding of dental floss 30 is extremely important, as fraying and shredding are the most frequently encountered consumer complaints about floss.

Traditionally, floss consists of continuous fibers coated with wax containing additives such as flavors, sweeteners and one or more active ingredients. The microcrystalline wax that is currently used holds the fibers together and 5 facilitates the repetitive sliding motion of floss between teeth. Shear forces applied to floss during use lead to fraying and shredding of the fibers. Such fraying and shredding occur primarily because the stresses applied to the floss during use tend to exceed the cohesive forces of 10 wax that help bond the fibers together.

There are two possible routes that can be adopted in order to overcome the floss shredding problem. These include making the floss from a monofilament of suitable size or a 15 "pseudo-monofilament", i.e., bonding the plurality of filaments in a multifilament structure such that they are adhered together and function like a monofilament. Pseudo-monofilament dental flosses are very similar to fiber-reinforced composites, in which fibers are 20 impregnated with polymer matrices, which can be thermoset or thermoplastic in nature.

Composites are matrixes, e.g., a matrix comprising a polymeric resin, that are reinforced with another material. For fiber composites, the reinforcing material 25 is fibers. The composite may be used as a dental floss. A study of the available composite manufacturing techniques leads to the understanding that two important stages exist. In stage one, the fibers and binder polymers are brought into intimate contact. In stage two, 30 heat and pressure are applied in order to impregnate and consolidate the components. Stage one is crucial, as it brings the matrix polymer and fiber in closer proximity to each other, thereby minimizing the flow length required during consolidation. This first stage is what makes

processing techniques used for thermoplastic composites different from thermoset composites, due to their higher viscosities.

- 5 It is known to overcome this problem by trying to reduce the viscosity of the resin in order to achieve rapid impregnation of the reinforcing fibers. This is called a pre-impregnation processes.
- 10 Solvent or solution impregnation has been used primarily where the high viscosity of the matrix material is reduced using solvents or plasticizers by dissolving the polymer in the solvent. The fibers are then made to pass through a dip bath filled with the solution of matrix material.
- 15 The fibers are coated and the coated fibers are then passed through a series of dryers in order to remove the solvent, thereby providing the finished composite. The biggest disadvantage of such a process is environmental concerns regarding use of the solvent. In addition,
- 20 manufacturing speed is very low, and thus manufacturing costs are increased.

The process described above requires dissolving a coating material in a solvent prior to coating a substrate.

- 25 Dissolving the coating material may be undesirable, as there is an environmental concern over volatile organic compounds. Therefore, there is a need for a process of coating a substrate that does not require dissolving a coating in a solvent.

30

Powder impregnation is a more versatile process, as it will process both low and high viscosity resins as long as they can be obtained in the powder form, and the process is relatively simple.

Investigators at Georgia Tech have developed a system wherein glass fibers are spread using up to 8 Teflon* coated rollers after which the spread fibers have a powder 5 deposited thereon using a deposition system developed by Electrostatic Technology Incorporated (ETI). In this system, powder particles are charged and are then electrostatically deposited onto the glass fiber. The above-mentioned rollers were found, however, to cause 10 damage to the fiber and a transition was then made to a different fiber spreading technology known as pneumatic venturi spreading. This effort lead to the issuance of United States Patent No. 5,094,883. In the patent, the 15 importance of flexible fiber impregnation production was taught for applications such as braiding and weaving.

None of the dental floss patents of which the inventor is aware have applied a powder technology approach for coating the substrate fibers of the dental floss with 20 polymers. There is a continuing need for a process of coating substrates such as fibers and medical devices which does not require dissolving a coating in a solvent.

The present invention provides a process including: 25 providing a substrate; and electrostatically coating the substrate with at least one coating material. In another embodiment, the present invention provides a substrate coated by electrostatic impregnation. The invention utilizes electrostatic powder coating technology to coat a 30 substrate with materials such as waxes; thermoplastic polymers; additives such as spray-dried flavors and sweeteners; active ingredients such as sodium fluoride; abrasives; etc. This method can be used for coating any substrate including, but not limited to films, non-wovens,

monofilament fibers, multi-filament fibers, medical devices, hair, sutures, and metal devices as long as the coating materials are in a powder form. The preferred substrates are monofilament fibers, multi-filament fibers, 5 and medical devices. The coated fibers may be useful in applications such as, but not limited to, dental tapes and dental floss. The medical devices may be made of ceramics, metal alloys, or polymers. The medical devices may be useful in applications such as, but not limited to, 10 stents and polymer tubes such as catheters.

The approach for electrostatically coating floss adopted in the present invention is to prepare a pseudo-monofilament by using waxes or thermoplastic polymers to 15 adhere the fibers to each other before coating the resulting pseudo-monofilament with a desired coating composition which may include, for example, waxes, thermoplastic polymers, flavors and other additives. Some of the advantages of using waxes or thermoplastic polymers 20 as the coating materials are the ease of processability, toughness, durability, long shelf life, lack of cross-linking chemical reactions and relatively high manufacturing speed. The drawbacks, however, are very high melt viscosities (in the range of 10^4 poise) which 25 lead to challenges in the areas of total fiber wet-out, interface control and mass production.

In the present invention, wax or polymer powder impregnation has been chosen to accomplish the challenge 30 of bringing the fiber and matrix into contact by using an electrostatic deposition chamber. The technique has the ability to support continuous production of fibrous substrates which can then be integrated into a consolidation line that can use techniques such as

calendering, hot-gun heating, filament winding and hand lay-up to apply pressure and heat in order to produce the floss product. Once the fibers have been bonded together, wax and other additives can be applied to the bonded 5 substrate.

The physics of charging polymeric particles is not always easy to comprehend and is far from being completely tied to the chemistry of the polymer powder. However, a 10 considerable amount of research has been done in the area with regard to electrostatic spray guns in the painting and coating industry.

Powders acquire charge in two ways: tribocharging and 15 corona charging. Corona charging results when particles receive a charge from electrically charged air. Tribocharging occurs when powder, during transportation from a reservoir to the spray gun or coater bed, undergoes frictional contact with an unsymmetrical surface. This 20 unsymmetry could be due to velocity, temperature or chemical composition. The polarity and magnitude of the tribocharge depend on the nature of the powder, the travel velocity, and the nature of the contact tubing. Formulation chemistry of the polymer could also determine 25 the nature, i.e., positive or negative, of the acquired charge. Usually the charging is higher at lower transportation rates and decreases as transportation rates increase. On the other hand, corona charging occurs in the region between the corona glow and the substrate. For 30 particles beyond a certain size (>0.5 microns) field charging predominates. Below a size of 0.2 microns however, the charge diffuses through the air.

According to theory, the saturation charge per ball-shaped particle is directly proportional to the square of the particle radius and inversely proportional to its mass. The shape of the particle was found to not deviate in most 5 cases from the ball-shape and so approximations made with regards to the spherical nature of the particle in theoretical calculations are still fairly accurate. In regard to the sign (i.e., positive or negative) and magnitude of the charge the electron affinity of the 10 elements or functional groups bound to the carbon atoms and their stereometric arrangement in the macromolecule proves to be decisive.

A comparison shows that the increasing tendency to charge 15 negatively moves in accordance with the increasing work function of the electron [poly methyl(methacrylate), polyethylene, poly (vinylchloride), poly (tetrafluoroethylene)], while materials with the lowest work functions tend to charge positively (e.g., 20 polyamide). In spray gun applications, aerodynamic forces are responsible for transporting the powder towards the object and electrostatic forces are responsible once the powder is near the substrate in order to facilitate good wrapping of the coating around the fibers.

25

Suitable fibers to be used in the present invention include, but are not limited to, natural fibers such as cellulose, cellulosic fibers, and rayon; polyolefins such as polyethylene and polypropylene; polyesters such as 30 polycaprolactone ("PCL"), poly(ethylene terephthalate) ("PET"), poly(butylene terephthalate) ("PBT"), and Vectran (Trademark of Hoechst-Celanese); polyamides such as nylon 6, nylon 11, nylon 12, and nylon 6,6; poly(ether-amides)

such as, but not limited to, Pebax® 4033 SA and Pebax® 7233 SA (Trademark of Elf Atochem); poly(ether-esters) such as, but not limited to, Hytrel® 4056 (Trademark of DuPont) and Riteflex® (Trademark of Hoechst-Celanese); 5 fluorinated polymers such as poly(vinylidene fluoride) and poly(tetrafluoroéthylène); and combinations thereof, including bicomponent fibers, which may be core-sheath fibers. Texturized fibers may also be used.

The bicomponent fibers may have cross-sectional shapes 10 such as round; trilobal; cross; and others known in the art. The core-sheath bicomponent fibers are typically made such that the sheath of the fibers utilizes a lower melting point polymer than the core polymer. Suitable polymers for the core include polyamides such as, but not 15 limited to, nylon 6, nylon 11, nylon 12, and nylon 6,6; polyesters such as, but not limited to, PET and PBT; poly(ether-amides) such as, but not limited to, Pebax® 4033 SA and Pebax® 7233 SA (Trademark of Elf Atochem); poly(ether-esters) such as, but not limited to, Hytrel® 20 4056 (Trademark of DuPont) and Riteflex® (Trademark of Hoechst-Celanese); polyolefins such as, but not limited to, polypropylene and polyethylene; and fluorinated polymers, such as, but not limited to, poly(vinylidene fluoride); and mixtures thereof. Nylon 6 and 25 polypropylene are preferred.

Suitable polymers for the sheath include polyolefins such as, but not limited to, polyethylene ("PE") and polypropylene; polyesters such as, but not limited to, 30 PCL; poly(ether-amides) such as, but not limited to, Pebax® 4033 SA and Pebax® 7233 SA (Trademark of Elf Atochem); poly(ether-esters) such as, but not limited to,

Hytrel® 4056 (Trademark of DuPont) and Riteflex® (Trademark of Hoechst-Celanese); elastomers made from polyolefins, for example Engage® elastomers (Trademark of DuPont-Dow); poly(ether urethanes) such as, but not limited to, Estane® (Trademark of BF Goodrich); poly(ester urethanes) such as, but not limited to, Estane® (Trademark of BF Goodrich); Kraton® polymers (Trademark of Shell Chemical Company) such as, but not limited to poly(styrene-ethylene/butylene-styrene); and poly(vinylidene fluoride) copolymers, such as, but not limited to, KynarFlex® 2800, (Trademark of Elf Atochem). Pebax® polymers, polyethylene, and PCL are preferred.

The ratio of the two components of the core-sheath fibers can be varied. All ratios used herein are based on volume percents. The ratio may range from about 10 percent core and about 90 percent sheath to about 90 percent core and about 10 percent sheath, preferably from about 20 percent core and about 80 percent sheath to about 80 percent core and about 20 percent sheath, more preferably from about 30 percent core and about 70 percent sheath to about 70 percent core and about 30 percent sheath. The sheaths of the bicomponent fibers may be fused prior to electrostatic coating.

25

The substrates are electrostatically coated with at least one coating composition. Suitable first coatings to be used in the present invention include, but are not limited to, poly(ethylene oxide); poly(ethylene glycol); hydroxyethyl cellulose; hydroxypropyl cellulose; polyethylene; waxes such as microcrystalline wax; polyvinylidene fluoride and polycaprolactone.

Suitable second coatings to be used in the present invention include, but are not limited to, poly(ethylene oxide), poly(ethylene glycol), hydroxyethyl cellulose, hydroxypropyl cellulose, polyethylene, waxes such as 5 microcrystalline wax, and polycaprolactone. The coatings may contain flavors, such as, but not limited to, natural and synthetic flavor oils, such as mint and cinnamon. The flavor oils may be used as is, or may be encapsulated in or supported on a carrier such as starch or modified 10 starch.

The process of the invention may also be useful for orienting short bicomponent fibers perpendicular to the axis of a substrate, then fusing the short bicomponent 15 fibers to the substrate. The length of the short bicomponent fibers may range from 1 mm to 5 mm. After the short bicomponent fibers are oriented onto the substrate by electrostatic impregnation, they are fused to the substrate by heat at a temperature appropriate to melt the 20 outer surface of the bicomponent fiber.

Additional materials that may be included in the coatings include, but are not limited to, sweeteners such as bulk sweeteners, including sorbitol and mannitol, and intense 25 sweeteners including aspartame and sodium saccharin, as taught by European Patent Application EP 919,208, hereby incorporated by reference for the disclosure relating to waxes and sweeteners; abrasives, such as silica; dentrifices; chemotherapeutic agents; cleaners; and 30 whiteners. Examples of suitable additives are disclosed in United States Patent No. 5,908,039, the disclosure of which is hereby incorporated by reference. Any of the foregoing materials may be used in encapsulated form.

The amount of wax, flavor, and other additives typically coated on fibers to make floss is known in the art. Typically, the coating composition is added at from 2 weight percent to 60 weight percent, based on the weight 5 of the fibers.

Suitable medical devices to be used in the present invention include, but are not limited to, medical devices made from ceramics, metal alloys, and polymers, such as 10 stents, and polymer tubes such as catheters. The medical devices may be coated with the same waxes and in the same manner as mentioned above.

Many factors were important in determining how to prepare 15 examples of the present invention. The selection of materials was based on a number of chemical and physical parameters of the polymer and the fiber. Among these, the melting point (T_m) and degradation temperature (T_d) are important as they determine issues regarding material 20 selection and processing conditions. The chemical structure of the polymers will influence the adhesion of the deposited material to the substrate fibers.

Processing conditions further involve operating furnace 25 temperature, residence time in the furnace (which influences line speed) and powder concentrations. Physical parameters such as powder particle size (mean and distribution), density, melt viscosity, electrical conductivity, etc. are among the important ones that 30 determine the deposition characteristics. Crystallization kinetics of the binder polymer also must be considered as it has an influence on the morphology/structure of the polymer once it is cooled down from the melt during the consolidation stage. Coating polymers and fibers were

chosen for the examples considering all the factors, and are summarized with their properties in Tables 1A and 1B.

Table 1A

<u>Polymer</u>	<u>Producer</u>	<u>Particl e Size</u>	<u>T_M</u> (°C)	<u>T_D</u> (°C)
Polycaprolactone (Tone 767)	Union Carbide	60	60	343
Polyethylene Oxide (WSR-N-10-100- Reg)	Union Carbide	100	65	250
Polyethylene Oxide (WSR-N-80-100- NF)	Union Carbide	100	65	250
Nylon 11 (Besno)	Elf Atochem	NT	185	436
Poly(vinylidene Fluoride) (Kynar 711)	Elf Atochem	200	165	374
PVDF Copolymer (Kynar-Flex 2801)	Elf Atochem	200	145	363
Poly(ether- amide) (Pebax 4033-SA)	Elf Atochem	NT	160	232
High Density Polyethylene (HDPE GHR 8110)	Hoechst- Celanese	120	130	NT
Hydroxypropyl cellulose (Klucel [EXS Pharm.])	Hercules	100	190	350

5 (particle size is in mesh)

NT = not tested

Table 1B

<u>Fiber Type</u>	<u>Construction</u>	<u>Denier</u>	<u>Tenacity</u> (g/d)	<u>T_M</u> (°C)	<u>T_D</u> (°C)
Nylon 6,6	Untwisted Air- Entangled 3 dpf	630	8	255	442
Nylon 6	Untwisted Air- Entangled 3 dpf	1400	5	220	446
Polypropylene	Untwisted Air- Entangled 3 dpf	630	8	162	301
Teflon	Monofilament	1200	5	N/A	NT
Polyester (Vectran)	Untwisted 5 dpf	400	23	331	NT

dpf = denier/filament

The powder coating process is shown schematically in
5 Figure 1. The coating line consists of a feed spool,
grounding unit, electrostatic coater, furnace/oven and
take-up winder. The feed spool is mounted on a metallic
post that facilitates easy unwind of the fiber during
operation. The fiber is then made to pass through an
10 eyelet that is connected to a ground source to allow for
grounding of the fiber. The fiber then passes through a
tensioning device made of ceramic rods that can be
adjusted depending on the desired packing conditions of
the powder in the fiber matrix. The fiber then passes
15 through openings provided at each end of the coater.

A B-60 coater (available from Electrostatic Technology
Incorporated, a subsidiary of Nordson Corporation,
Branford, CT) was used to conduct the processes set forth
20 in the examples. The coater comprises a powder feeder
connected to it that feeds powder, which is held in a

hopper, into the rear end of the bed. A photohelic gauge placed in the bed measures the level of powder in the bed and ensures that a constant level of powder is maintained during the coating operation. A refrigerator unit ensures 5 the delivery of clean, moisture and oil free air into the plenum of the coater, and a powder collector provides for collection of powder that is vacuumed out of the bed during operation. The coater also comprises a fire detection device and a vortex tube to assist in the 10 formation of the powder cloud.

The powder coated fiber exits the deposition chamber and then passes through a furnace. A suitable furnace and is made by Lindberg and Glenro. The furnace has a spilt lid 15 that is made to open and close via air actuated arms. The fiber then moves through an eyelet onto a core spool using a Leesona rewinder. The line speed can be controlled by the Leesona rewinder and is measured using a digital tachometer.

20

A layer of polymer powder was fluidized in the bed, which was constructed of a plastic material. The powder was fluidized by air that was charged negatively in the plenum of the coater. The air was kept clean and free of 25 moisture and oil by passing it through a refrigeration unit. The air was then made to pass through the air plenum and encountered a mesh of electrodes that were charged by a highly negative direct current supply.

30 Negative coronas used as negatively charged powder particles tend to deposit more uniformly and efficiently than positively charged ones because of their relative resistance to electric breakdown. The electrons generated in the glow region of the negative corona quickly attach

themselves to electronegative gas molecules such as oxygen in the air to form negative ions. The negatively charged air then passed through a porous ceramic or plastic bed to contact the powder. Field charging or ion bombardment 5 then transferred some of the ions to the powder particles thereby generating an aerated, negatively charged cloud of powder in the bed. This indirect charging of the powder by the air along with the separation of the charging mesh from the powder makes the fluidized bed process different 10 from other powder techniques such as powder spray gun and triboelectric charge guns. The substrate, which in the present case was the fiber, was then grounded in order to generate a sufficient potential difference to facilitate electrostatic deposition.

15

The amount of polymer binder added to the fibers was the main dependent variable for the design of experiments. The powder coating variables that could have an impact on the add-on are bed fluidization flow rate ("m³/sec"), 20 electrostatic voltage (kilovolts, "kV") and line speed (meters per minute). The other variables that can be controlled on the coater are bed air pressure, vortex flow rate and vortex air pressure. It was determined that none of these latter variables influenced the add-on 25 significantly.

The vortex tube settings were changed to create the powder cloud and once the cloud was generated, the settings were kept constant throughout the experiment. The bed agitator 30 was turned on whenever needed, but was avoided as much as possible, as it was seen to reduce the add-on due to additional turbulence in the powder cloud. In general, wherever possible, a two parameter, three level design of

experiments was conducted. The variables were varied within the following limits:

5 Bed fluidization flow rate (m³/sec): Low (0.006);
Medium (0.010); and High (0.014)

Electrostatic voltage (kV): Low (0 or no voltage);
Medium (20.2); and High (40.4)

10 It was important to determine the furnace temperature in order to optimize the polymer fusion and bonding on the fiber. A visual experiment was carried out on nylon 6 fibers and polyethylene oxide binder. The residence time in the furnace for a line speed of 10 m/min was calculated
15 based on the length of the furnace (approximately 1 meter) to be 6.3 seconds.

20 Fiber was coated at a bed flow rate of 0.014 m³/sec and electrostatic voltage of 40.2 kV and was then inserted into the furnace at a selected temperature and held there for 6.3 seconds. Optical microscopy was performed to check the melt morphology of the polymer on the fibers. The furnace temperature was then raised and a freshly coated sample was exposed to the temperature. This was
25 done until a temperature was reached at which melt occurred. It was observed that a good coating was formed on the fibers at a furnace temperature of 265°C.

30 The procedure described above was utilized to determine the furnace temperature for a given set of materials and process conditions. The residence time in the oven can be increased by using longer ovens or by wrapping the coated yarn in grooves of metallic rollers mounted on pulleys in the heated section of shorter ovens.

Example 1 - Polyamide Multi-filaments/Water Soluble
Polymer Coating Systems

5 (a) Nylon 6 Multi-filaments/Poly(Ethylene Oxide)

Poly(ethylene oxide) [PEO] N-80 grade was used to coat a nylon 6 multi-filament structure comprising 467 filaments, each filament having a denier of 3. Tables 2 and 3 show 10 the add-on (%) of PEO as a function of electrostatic voltage and bed flow rate, respectively, at an oven temperature of 260°C and a line speed of 16 m/min. The add-on increases with increasing voltage but decreases beyond a voltage value of 40 kV. On the other hand, the 15 trend with respect to bed flow rate indicates a strong monotonic dependence where the add-on increases with increasing flow rate. PEO powder was uniformly dispersed and fused on the nylon 6 fibers as was observed by scanning electron microscopy.

20

Table 2

<u>Bed Flow Rate (m³/sec)</u>	<u>Electrostatic Voltage (KV)</u>	<u>Add On (%)</u>
0.006	0	1.73
0.006	20	9.8
0.006	40.4	10.2
0.006	60.2	1
0.010	0	4.17
0.010	20	22.64
0.010	40.4	24.14
0.010	60.2	3.75
0.014	0	7.14
0.014	20	28.57
0.014	40.4	36.21
0.014	60.2	5
0.014	80.4	5.35

Table 3

<u>Electrostatic Voltage (KV)</u>	<u>Bed Flow Rate m³/sec</u>	<u>Add On (%)</u>
0	0.006	1.73
0	0.010	4.17
0	0.014	7.14
20.6	0.006	9.8
20.6	0.010	22.64
20.6	0.014	28.57
40.3	0.006	10.32
40.3	0.010	24.14
40.3	0.014	36.21
60.2	0.006	1
60.2	0.010	3.75
60.2	0.014	5
80.4	0.014	5.35

(b) Nylon 6,6 Multi-filaments/Poly(ethylene Oxide):

PEO N-80 was used to coat a nylon 6,6 multi-filament structure comprising 210 filaments, each having a denier of 3, at a furnace temperature of 237°C. Tables 4 and 5 show the add-on of PEO as a function of voltage and flow rate at line speeds of 11 m/min and 17 m/min, respectively. As seen from the Tables, regardless of the line speed, the add-on is very sensitive to bed flow rate, with increasing flow rates resulting in higher add-on. The trend with voltage is the same as in the previous case with the existence of a saturation voltage beyond which increasing voltage results in a drop in add-on. The saturation voltage in the present case appeared to be 40 kV. The add-on also seemed to be higher at 17 m/min as compared to 11m/min, but higher speeds did not support this trend. Scanning electron micrographs showed that PEO was uniformly coated on the nylon 6,6 fibers.

20

Table 4

<u>Voltage</u>	<u>Add-On (%)</u>	
	<u>0.006 m³/sec</u>	<u>0.014 m³/sec</u>
0	5.13	19.23
20	7.14	22.22
40	3.33	15.69
60	3	15.38

Table 5

<u>Voltage</u>	<u>Add-On (%)</u>	
	<u>0.006 m³/sec</u>	<u>0.014 m³/sec</u>
0	2.5	17.07
20	3.64	24.1
40	3	26.67
60	1	22.73
80	0.6	23

A study was conducted after coating nylon 6,6 with 20% PEO at an oven temperature of 235°C, voltage of 30 kV, flow rate of 0.013 m³/sec, bed pressure of 45 psi, vortex pressure of 20 psi and line speed of 11 m/min. Table 6 summarizes the results of a consumer test of PEO coated nylon 6,6 floss. The data (percent of people who indicate the floss passes each test) shows that the PEO coated floss performed well, particularly at ease of sliding between teeth and being gentle on the gums.

Table 6

<u>Property</u>	<u>Percent Pass</u>
Sliding easily between teeth	63
Being gentle to the gums	55
Cleaning effectively between all teeth	24

(c) Nylon 6,6 Multi-filaments/Hydroxypropyl Cellulose:

Nylon 6,6 was coated with hydroxypropyl cellulose. Tables 7 (line speed of 11 m/min, oven = 235°C) and 8 (line speed of 16 m/min, oven = 255°C) show that the add-on of hydroxypropyl cellulose ("HPC") was dependent on the bed flow rate which was the dominant factor. HPC powder possessed large amounts of charge as received from the supplier. The powder floated and tended to melt onto the fiber once deposited.

Table 7

<u>Voltage</u> <u>(kV)</u>	<u>Add-on (%) at 0.006</u> <u>m³/sec</u>	<u>Add-on (%) at 0.014</u> <u>m³/sec</u>
0	0	9.09
40	2	11.76
80	1.2	12.5

15

Table 8

<u>Voltage</u> <u>(kV)</u>	<u>Add-on (%) at 0.006</u> <u>m³/sec</u>	<u>Add-on (%) at 0.014</u> <u>m³/sec</u>
0	0	14.81
40	3.57	18.18
80	4	15

Example 2 - Polyester Multi-filaments/Water Insoluble Polymer Coating Systems

(a) Vectran Multi-filaments/High Density Polyethylene:

5

Vectran is a high performance/ high temperature fiber made from liquid crystalline polyester and is commercially available from Hoechst-Celanese. A furnace temperature of 310°C was used to melt the water-insoluble polyethylene 10 coating material onto the Vectran® fiber at a line speed of 17 m/min. Table 9 shows that in order to achieve higher add-ons of polyethylene, high flow and voltage was essential.

15

Table 9

<u>Voltage (KV)</u>	<u>Add On (%)</u>
0	9.35
20.2	8.38
40.3	8.1
60.1	14.8
80.5	55

The flow rate required to achieve the above add-ons was 0.014 m³/sec.

20

Example 3 - Fluoropolymer Monofilament/Water Soluble Polymer Coating Systems

(a) Poly(tetrafluoroethylene) ("PTFE")

Monofilament/Poly(ethylene oxide) :

25

PTFE monofilament was coated with a non-wax water-soluble polymer, i.e., poly(ethylene oxide), to increase the coefficient of friction. The PTFE monofilament was generally rectangular in transverse cross-section, with a

width of about 2-3 mm and a thickness of about 0.08 - 0.13 mm. The temperature was 320°C at a line speed of 27 m/min. Table 10 shows the add-on of PEO on the PTFE monofilament tape, and it was observed that beyond 20 kV voltage the 5 add-on drops as was seen earlier. This shows that the trend of add-on versus voltage is independent of substrate geometry, i.e., cylindrical versus flat fibers. PTFE was also successfully coated with a mixture of multiple powders such as PEO, spray-dried peppermint flavor and 10 sodium saccharin.

Table 10

<u>Voltage</u> <u>(kV)</u>	<u>Add-On (%)</u>		
	<u>0.006 m³/sec</u>	<u>0.010 m³/sec</u>	<u>0.014 m³/sec</u>
0	10.26	10.22	NT
21.0	18.39	18.7	21.62
40.2	10.58	11.9	NT
60.2	7.5	9.1	NT
80.4	9.1	11.2	NT

Tables 11A and 11B summarize all the coating experiments 15 that were conducted on different classes of fibers using different polymer coating systems and the preferred conditions for each set of materials. The main variables were bed pressure (0.006 to 0.014 m³/sec), voltage (0 to 80- kV) and line speed (6 to 64 m/min for Nylon 6/PEO N80, 20 and 11 to 17 m/min for the rest of the materials tested). The furnace temperature was changed based on the fiber/polymer combination. If desired, further consolidation can be carried out on the powder-coated fibers in order to prepare a thoroughly impregnated 25 product. A combination of heat and pressure can be used

to drive the polymer inside the fibers that will produce a void free product. Using the viscous flow of the polymer through the fibrous network and the elastic deformation of the fiber network, a consolidation model can be developed.

5

Table 11A

		Preferred Conditions			
<u>Fiber</u>	<u>Polymer</u>	<u>Pressu re</u>	<u>Voltage</u>	<u>Speed</u>	<u>Comments</u>
Nylon 6	PEO N80	0.014	40.4 kV	11 m/min	Oven = 260°C Binder = 1.75% to 31%
Nylon 6,6	PEO N80	0.014	20.2 kV	11 m/min	Oven = 237°C Binder = 1.5% to 27% Reduced add on due to turbulence
	PEG 8000				Larger particle size, High density / poor floatation, Low molecular weight, good melt
	Klucel	0.014	40.2 kV	17 m/min	Powder was very fine, Extremely cohesive, Agitator had to be used, Feed hopper could not be used

Table 11B

<u>Fiber</u>	<u>Polymer</u>	<u>Preferred Conditions</u>			<u>Comments</u>
		<u>Pressure</u> e	<u>Voltage</u> e	<u>Speed</u>	
Nylon 6,6	Kynar 711 Kynar Flex PCL				Kynar powders - high densities, fine particle size, PCL did not charge due to its size, grinding down was not possible
Polypropylene	PEO N80	0.010	20.2 kV	11 m/min	Fiber performed very well, low melting point made coating difficult
Vectran	HDPE	0.014	80.2 kV	17 m/min	Powder was extremely heavy, oven = 310°C
Teflon	HDPE PEO N80, flavor, saccharin	0.014 0.006	80.2 kV 20.2 kV	17 m/min 27 m/min	Powder was extremely heavy, oven = 310°C, good melt but poor adhesion Excellent floatation, oven = 320°C, increased blend ratio

Example 4 - Selection Process for Polymer Powders Based on Powder Aeration

It is necessary to understand powder aeration to predict 5 the coating performance of polymer powders. In order to understand this quantitatively, experiments were carried out on several powders to obtain their bulk and tapped densities. In this experiment, a clean graduated cylinder was taken and weighed (W_1). 50 cc. of powder was then 10 poured into the cylinder and weighed again (W_2). Weights were recorded to 5 decimal precision. The aerated/bulk density was then computed using equation (1).

$$(1) \quad \rho_b = \frac{W_2 - W_1}{50} \quad , \text{ where}$$

15 ρ_b is the bulk density in grams/cc.

In order to obtain the tapped density the same measurements were made with a minor difference. After the powder was poured into the cylinder, it was tapped 50 20 times in order to compact it and powder was added to maintain a 50ml volume. The cylinder was then weighed (W_3). Equation 1 was used after substituting W_3 for W_2 , and the tapped density was obtained in grams/cc. Table 12 lists the bulk and tapped densities of the different 25 polymers. The fused density in the table represents the density of a block of material.

Table 12

<u>Polymer</u>	<u>Bulk Density</u> (g/cc)	<u>Tapped Density</u> (g/cc)	<u>Fused</u> <u>Density (g/cc)</u>
Polyethylene Oxide WSR-N-10-100-Reg WSR-N-80-100-NF	0.46-0.50	0.54	1.2
Polycaprolactone 767	0.52	0.61	1.15
Polyethylene Glycol	0.54	0.82	1.03
Polyethylene GHR 8110	0.47	0.51	0.95
HPC (Klucel)	0.32	0.39	1.16

5 The differential densities (fused density - tapped density) of the polymer particles were plotted (log-log) as a function of mean particle size (in microns). It appears that the relationship of these polymer properties is critical to the polymer's fluidization properties, such 10 as aeratability, cohesiveness, sand-like properties, and spoutability. The data can be used as a guide in selecting powders for fluidized-bed coating applications. For example, PEO and HPC are more aeratable than Polyethylene glycol and Polyethylene GHR 8110 (a high 15 density polyethylene). Therefore, PEO and HPC are more easily fluidized.

I Claim:

- 1) A process comprising:
 - providing a substrate selected from the group consisting of films, non-wovens, monofilament fiber, multi-filament fibers, medical devices, hair, sutures, and metal devices;
 - and electrostatically coating the substrate with at least one coating material.
- 10 2) The process of claim 1, wherein the substrate comprises at least one fiber.
- 15 3) The process of claim 2, wherein the fiber comprises a material selected from the group consisting of cellulosics; polyamides; polyesters; polyolefins; poly(ether-amides); poly(ether-esters); fluorinated polymers; copolymers thereof; and blends thereof.
- 20 4) The process of claim 3, wherein the polyamide is selected from the group consisting of nylon 6, nylon 11, nylon 12, and nylon 6,6;
 - the polyester is selected from the group consisting of polycaprolactone; poly(ethylene terephthalate);
 - 25 poly(butylene terephthalate) and Vectran;
 - the polyolefin is selected from the group consisting of polypropylene and polyethylene;
 - the poly(ether-amide) is selected from the group consisting of Pebax® 4033 SA polymer and Pebax® 7233 SA polymer;
 - 30 the poly(ether-ester) is selected from the group consisting of Hytrel® 4056 polymer and Riteflex® polymer; and

the fluorinated polymer is poly(vinylidene fluoride).

- 5) The process of claim 4, wherein the coating is selected from the group consisting of poly(ethylene oxide), poly(ethylene glycol), hydroxyethyl cellulose, hydroxypropyl cellulose, polyethylene, wax, elastomers, polycaprolactone and combinations thereof.
- 6) The process of claim 5, wherein a second coating is 10 electrostatically applied to the substrate.
- 7) The process of claim 6, wherein the second coating is selected from the group consisting of poly(ethylene oxide), poly(ethylene glycol), hydroxyethyl cellulose, 15 hydroxypropyl cellulose, polyethylene, wax, elastomers, polycaprolactone and combinations thereof.
- 8) An article comprising:
a substrate selected from the group consisting of 20 non-wovens, monofilament fibers, multi-filament fibers, films, medical devices, hair, sutures, and metal devices; wherein the substrate is coated by the process of claim 1.
- 9) The article of claim 8, wherein the substrate 25 comprises at least one fiber.
- 10) The article of claim 9, wherein the fiber is made from a material selected from the group consisting of a polyamide; a polyester; a polyolefin; a poly(ether-amide); 30 a poly(ether-ester); a fluorinated polymer; copolymers thereof; and combinations thereof.

11) The article of claim 10, wherein

the polyamide is selected from the group consisting of nylon 6, nylon 11, nylon 12, and nylon 6,6;

5 the polyester is selected from the group consisting of polycaprolactone; poly(ethylene terephthalate); poly(butylene terephthalate), and Vectran;

the polyolefin is selected from the group consisting of polypropylene and polyethylene;

10 the poly(ether-amide) is selected from the group consisting of Pebax® 4033 SA polymer and Pebax® 7233 SA polymer;

the poly(ether-ester) is selected from the group consisting of Hytrel® 4056 polymer and Riteflex® polymer; and

15 the fluorinated polymer is poly(vinylidene fluoride).

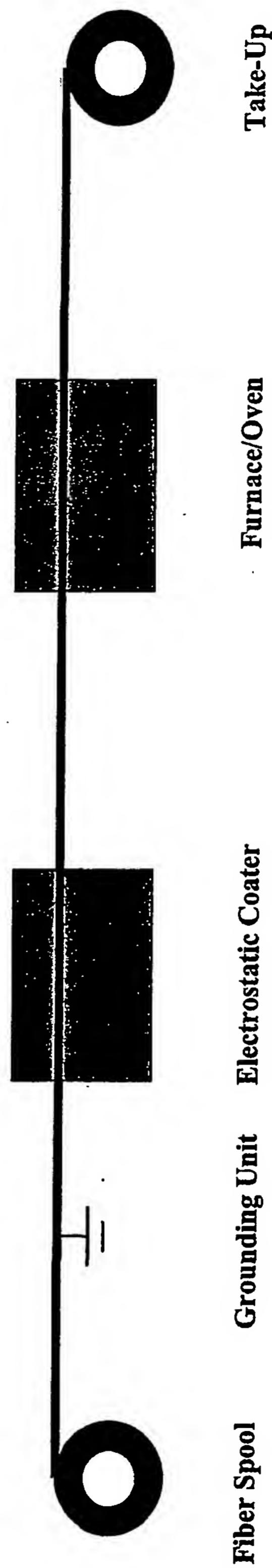
12) The article of claim 11, wherein the fiber further comprises a second coating.

20 13) The article of claim 12, wherein the first coating is selected from the group consisting of poly(ethylene oxide), poly(ethylene glycol), hydroxyethyl cellulose, hydroxypropyl cellulose, polyethylene, wax, elastomers, polycaprolactone and combinations thereof.

25

14) The article of claim 13, wherein the second coating is selected from the group consisting of poly(ethylene oxide), poly(ethylene glycol), hydroxyethyl cellulose, hydroxypropyl cellulose, polyethylene, wax, elastomers, 30 polycaprolactone and combinations thereof.

Figure 1



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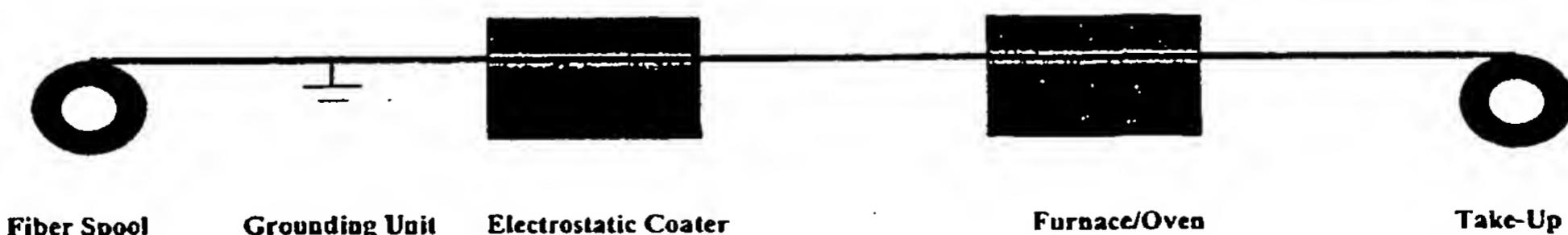
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(72) Inventor: **DAVE, Vipul, Bhupendra**; 20 Francis Drive, Belle Mead, NJ 08502 (US).

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(74) Agents: **JOHNSON, Philip S. et al.**; Johnson & Johnson, One Johnson & Johnson Plaza, New Brunswick, NJ 08933-7003 (US).

(54) Title: ELECTROSTATIC IMPREGNATION OF POWDERS ON SUBSTRATES

WO 02/02245 A3

Fiber Spool

Grounding Unit

Electrostatic Coater

Furnace/Oven

Take-Up

(57) Abstract: Electrostatic Impregnation Of Powders On Substrates The use of electrostatic impregnation to load materials such as binders and flavors onto substrates such as fibers and medical devices is disclosed. Substrates loaded with materials such as binders and flavors, wherein the materials are loaded on the substrates via electrostatic impregnation are also disclosed.

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International Application No

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A. CLASSIFICATION OF SUBJECT MATTER
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According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

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X	GB 2 190 927 A (SUCKER & FRANZ MUELLER GMBH) 2 December 1987 (1987-12-02) the whole document ---	1-5,8-11
X	US 4 345 907 A (WEGELE GUENTHER ET AL) 24 August 1982 (1982-08-24) the whole document ---	1-5,8-11



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Patent family members are listed in annex.

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